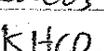
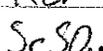
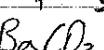
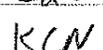
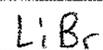
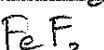
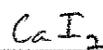
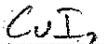
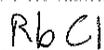


1. Name 3 chemical species isoelectronic with Ne.

2. Name the following ions:



3. Name the following ionic compounds:



4. Provide chemical formulas for the following:

Zinc(II) chloride

Magnesium sulfate

Lithium iodide

Nickel(II) chloride

Cesium oxide

Potassium chlorate

5. Draw the Lewis dot structure of the following atoms or ions:



6. Order the following from lowest to highest ionization energy:

